

Synthesis, Spectral Studies of Some New Chalcone and Schiff Base Derivatives Derived from Cyclohexanone and Molecular Docking and Biological Activity Studies

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Abstract

Through this study, a new series of novel cyclohexanone derivatives were synthesized as a Schiff Bases or chalcones were prepared by depending on the intermediate compound (Cyclohexanone) all new derivatives were characterized that was prepared by Fourier Transform Infrared Spectroscopy (FT-IR) Analysis, ¹HNMR spectroscopic data, and the ¹³C NMR spectrum, some the newly prepared derivatives which are important in the field of medical chemistry which were studied as anti-bacterial and anti fungal.

Keywords: Cyclohexanone, Schiff Bases, Chalcone, Anti-inflammatory Agent, Molecular Docking.

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INTRODUCTION

Cyclohexanone a cyclic compound containing six carbon atoms and one oxygen atom with the chemical formula CO(CH₂)₅

It is an important industrial compound useful as a raw material in organic synthesis⁽¹⁾, known as (Pimelic ketone, Keto-hexa methylene, Oxocyclohexane),⁽²⁾ it appears as a colorless to pale yellow liquid with a slight smell, less dense than water⁽³⁾. Cyclohexanone is an important intermediate in the production of ε-caprolactam which is used in the manufacture of Nylon-6 and polyamide resins,⁽⁴⁾ additional uses include paint and varnish removers, degreasing of metals and spot remover⁽⁵⁾.

Schiff Bases this name is given to organic compounds that contain in their composition azomethine as an active group⁽⁶⁾, here the carbonyl group has been replaced by an imine or azomethine group⁽⁷⁾.

It has been called Schiff Bases attributed to its explorer the scientist chemical Nobel Prize Winner German Hugo Schiff in 1864⁽⁸⁾. The compounds produced by the condensation of aldehydes with amines are called Aldimines but the compounds that are produced by condensing ketones with primary amines are called Ketimines⁽⁹⁾. Schiff bases are promising biologically interesting compounds having a wide range of pharmaceutical properties including antipyretic, anti-inflammatory, and antimicrobial activities⁽¹⁰⁾.

Chalcone named (α,β-unsaturated carbonyl compounds)

because it contains effective group $\left(\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}-\text{C}=\text{C}- \\ \text{H} \quad \text{H} \end{array} \right)$,⁽¹¹⁾ the presence of this group in the composition of chalcones makes

it possess a wide range of biological activities, and therefore it has an important place in biochemistry and medicine, due to its therapeutic properties that fight cancer, viruses, infections and combat⁽¹²⁾ HIV. the name chalcones was given by the two scholars Tombor, Kostanecki⁽¹³⁾, the compounds in which the main part of its plant structure is chalcone possesses biological and pharmacological activity, including anti-inflammatory, analgesic, antitumor, antimalarial, antihistamine, anti-oxidant, and antiprotozoal⁽¹⁴⁾.

EXPERIMENTAL

2-1-Materials and methods

All the used chemicals were obtained from commercial sources, with a purity range of 95-98%, that were used as received (without further purification). Melting points of all synthesized compounds were measured in open capillary tubes in a Gallen-Kamp MFB-600 melting point apparatus. FT-IR spectra measurements were recorded using FT-IR-8400S-Shimadizu spectrophotometer. ¹H-NMR and ¹³C-NMR spectra were recorded on VARIAN-INOVA 500MHZ spectrophotometer (Germany), CDCl₃ and DMSO-d₆ were used as solvents, and tetramethylsilane TMS as internal standard.

2-2-General Procedure for Schiff bases derivatives synthesis (3,5,7,9)⁽¹⁵⁾

In round bottle flask 250 ml, with a condenser and magnetic stirrer, (10mmol) of the compound (cyclohexanone) was added to (10mmol) of the compound (ethanol amine, 4-amino phenol, 3-amino phenol, thiosemicarbazide) respectively in

(30-40ml) of ethanol absolute as a solvent, and added three to four drops of glacial acetic acid as a catalytic, then mixture was raised by reflux condensation for a period of (10-12) hours and at a temperature of (78-80) °C the reaction was followed up using a thin layer chromatography (TLC) technique in a mixture of (Benzene, Methanol) at a ration (4:1) as a mobile phase The solvent is removed by distillation under vacuum pressure to obtain the product.

2-3-Synthesis drivatives of chalcones (11,13) ⁽¹⁶⁾

In round bottle flask 250 ml, with a condenser and magnetic stirrer, (10mmol, 0.98gn) of

the compound (cyclohexanone) was added to (20mmol, 2.44gm) (20mmol, 1.92gm) from the compound (4-hydroxy benzaldehyde, furfural) respectively in 4ml of THF as a solvent, and added 0.5ml of concentrated Hydrochloric acid as a catalytic, were mixed and stirred at room temperature up to 2 hours.

Then the temperature was raised to (50-60)°C and stirring was continued for 10 hours until the reaction was completed. The reaction is monitored using mobile phase TLC (Benzene-Methanol) in a ratio (4:1). The crude product was washed out with cold water and ethanol absolute (1:1), then filtered. The second washed out with cold water

and ethanol absolute (2:3) up to pH (7-8) reached out, then filtered and dried, a solid Compounds were purified by recrystallization.

4-2. Synthesis derivatives of chalcones (15,17) ⁽¹⁷⁾

In round bottle flask 250 ml, with tow neck, cyclohexanone (10mmol, 0.98gm) and mixture of the aromatic aldehyde such as (4-bromo benzaldehyde (20mmol, 3.7 gm) or vanillin (20mmol, 3.034gm), were dissolved in (15ml) of ethanol absolute, stirred mixture for several minutes at ice bath.

Then slowly add 10mL of a 40% sodium hydroxide solution in water to the mixture at room temperature for (6-8) hours, the reaction was followed up using a thin layer chromatography (TLC) technique in a mixture of (Benzene, Methanol) at a ration (4:1) as a mobile phase, the product was filtered and washed with cold water and dried, the product was purified by recrystallized by ethanol absolute.

Table 1. Physical properties of compound (C3-C17)

Com p.	Chemical formula	Color	Solubility	M.P(° C)	Yield %	Rf
C3	C ₈ H ₁₅ N O	Black	EtOH	Syrup	82.15	0.53
C5	C ₁₂ H ₁₅ N O	Black	EtOH	155-158	85	0.42
C7	C ₁₂ H ₁₅ N O	Black	EtOH	Syrup	76.6	0.49
C9	C ₇ H ₁₃ N ₃ S	Pale yellow	EtOH	155-157	83	0.54
C11	C ₂₀ H ₁₈ O 3	Dark green	THF	260-262	88.46	0.49
C13	C ₁₆ H ₁₄ O 3	brown	THF	130-132	84.6	0.81
C15	C ₂₀ H ₁₆ B r ₂ O	Pale yellow	EtOH	250 d	78	0.48
C17	C ₂₂ H ₂₂ O 5	yellow	EtOH	166-168	72.18	0.67

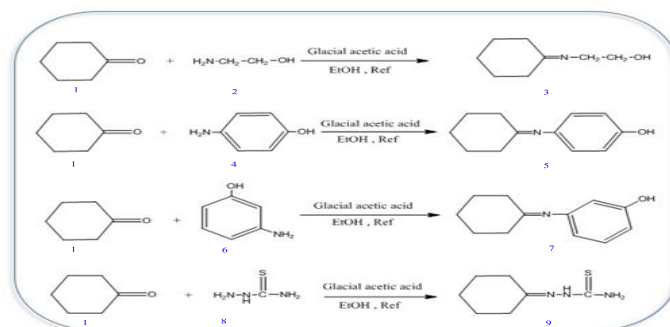


Figure 1. preparation of compound (3,5,7,8)

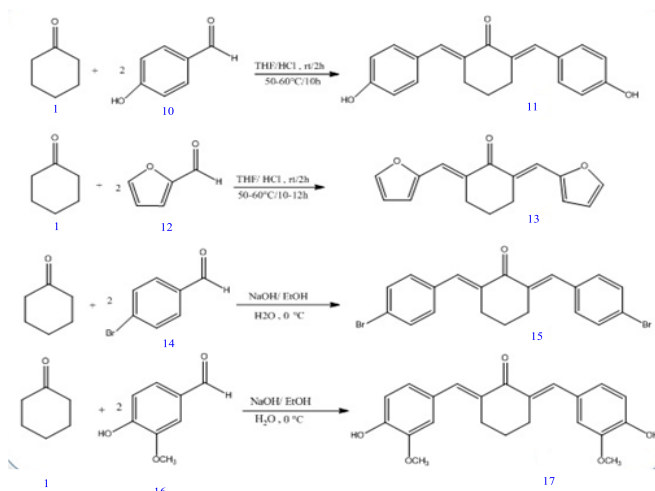


Figure (2) Synthesis compound (11,13,15,17)

RESULT AND DISCUSSION

Chemistry

All new compound have been identification by using FT-IR and $^1\text{H-NMR}$, $^{13}\text{CNMR}$ spectrum and used DMSO as a solvent.

Biological activity

Two different species of Gram-negative bacteria (*Escherichia coli*) and Gram-positive bacteria (*Staphylococcus aureus*) were investigated for biological activity in this work at a concentration of (100ppm) in media prepared for bacteria culture. Used a DMSO solvent with a volume of 5 mL with 0.0025 g from prepared derivatives.

Molecular docking

The molecular docking of the prepared compounds with the protein responsible for breast cancer was studied using AutoDock 4.2.6 program. This study predicted the presence of effective compounds towards the protein responsible for breast cancer.

3-1 The compound (**3**) 2-(cyclohexylidene amino)ethan-1-ol is black syrup, yield 82.15% The infrared spectrum of compound (**3**) showed band at (3263cm^{-1}) for (OH) alcohol, (2927cm^{-1}) for (C-H) aliphatic, (1566cm^{-1}) for (C=N) The $^1\text{H-NMR}$ compound (**3**) through the use (of DMSO-d₆) as a solvent. The proton signal (d,H1) appeared δ (1.77ppm), (s,CH₂) appeared δ (3.67ppm), (s,OH) appeared δ (6.95ppm), (s,H2) appeared δ (7.9ppm).

3-2 The compound (**5**) *N*-methyl cyclohexanimine phenol is black solid, M.P (155-159 yield 85.28%. The infrared spectrum of compound (**5**) showed band at (1558cm^{-1}) from (C=C) Aromatic, (2931cm^{-1}) for (C-H), (3332cm^{-1}) for (OH) phenol, (1667cm^{-1}) for (C=N). The $^1\text{H-NMR}$ (DMSO-d₆) compound(**5**) through the use (of DMSO-d₆) as a solvent. The proton signal (s,H1) appeared δ (1.63ppm), (d,H2) appeared δ (1.77ppm), (m,H3) appeared δ (2.35ppm) (m, Ar-H) appeared δ (4.8ppm), (s,OH) appeared δ (4.8ppm).). The $^{13}\text{CNMR}$ spectrum was used by (DMSO-d₆), showing the signal (C-Ar) appeared δ (115.67-121.1ppm), (C-N) appeared δ (148.67ppm), (C-OH) appeared δ (173.8ppm), (C=N) appeared δ (211.29ppm), (C1) appeared δ (25.73ppm), (C2) appeared δ (27.5ppm), (C3) appeared δ (30.83ppm).

3-3 The compound (**7**) 3-(cyclohexylidene amino) phenol is black syrup, yield 76.6% The infrared spectrum of compound (**7**) showed band at ($1402-1433\text{cm}^{-1}$) for (C=C) Aromatic, (2991cm^{-1}) for (C-H) aliphatic, (1606cm^{-1}) for (C=N), (1039cm^{-1}) for (C-O). The $^1\text{H-NMR}$ compound (**7**) through the use (of DMSO-d₆) as a solvent. The proton signal (t,H1) appeared δ (1.05ppm), (d,H2) appeared δ (1.51ppm), (s,H3) appeared δ (1.91ppm), (t,OH) appeared δ (5.3ppm), (m,H4) appeared δ (6.79ppm), (s,H5) appeared δ (7.75ppm) (t,H6) appeared δ (6.79ppm). The $^{13}\text{CNMR}$ spectrum was used by (DMSO-d₆), showing the signal (C1) appeared δ (20.46ppm),

(C2) appeared δ (25.56ppm), (C3) appeared δ (33ppm), (C-Ar) appeared δ (100.85-131.69ppm), (C-N) appeared δ (144.74ppm), (C-OH) appeared δ (149.87ppm), (C=N) appeared δ (158.52ppm).

3-4 The compound (**9**) 2-cyclohexylidene hydrazine-1-carbothioamide is solid light yellow, M.P (155-157), yield 83.5%. The infrared spectrum of compound (**9**) showed band at (2939cm^{-1}) for (C-H) aliphatic (1072cm^{-1}) for (C=S), (1581cm^{-1}) for (C=N), (3379cm^{-1}) for (N-H) primary, (1300cm^{-1}) for (N-H) bond.

The $^1\text{H-NMR}$ compound (**9**) through the use (of DMSO-d₆) as a solvent. The proton signal (s,NH) appeared δ (10.156ppm), (s,H1) appeared δ (1.051ppm), (s,NH2) appeared δ (7.96ppm), (t,H2) appeared δ (2.21ppm).). The $^{13}\text{CNMR}$ spectrum was used by (DMSO-d₆), showing the signal (C=S) appeared δ (179ppm), (C=N) appeared δ (157.45ppm), (C3) appeared δ (35.02ppm), (C2) appeared δ (27.66ppm), (C1) appeared δ (25.5ppm).

3-5 The compound (**11**) 2,6-bis(E)-4-hydroxybenzylidene)cyclohexan-1-one is dark green powder yield 88.46% M.P (260-263). The infrared spectrum of compound (**11**) showed band at ($1435-1589\text{cm}^{-1}$) for (C=C) aromatic, (1647cm^{-1}) for (C=O) ketone, (2931cm^{-1}) for (C-H), (3224cm^{-1}) for (OH) phenol.

The $^1\text{H-NMR}$ compound (**11**) through the use (of DMSO-d₆) as a solvent. The proton signal H1 appeared δ (1.5ppm), t, H2 appeared δ (1.72ppm), (d,CH=CH) appeared δ (6.83ppm), (mH-Ar) appeared δ (7.22-7.96ppm), (d,OH) appeared δ (5ppm). The $^{13}\text{CNMR}$ spectrum was used by (DMSO-d₆), showing the signal (C-Ar) appeared δ (116-136.27ppm), (C-OH) appeared δ (158.8ppm), (C=O) appeared δ (188.98ppm), (C1) appeared δ (28.44ppm), (C2) appeared δ (19.03ppm).

3-6 The compound (**13**) (2E, 6E)-2,6-bis(furan-2-ylmethylene)cyclohexan-1-one is solid brown, M.P(130-132), yield 84.38%.

The infrared spectrum of compound (**13**) showed band at (2939cm^{-1}) for (C-H) Aliphatic, (1643cm^{-1}) for (C=O), (1546cm^{-1}) for (C=C), (1284cm^{-1}) for (C-O), (1249cm^{-1}) for (C-O-C). The $^1\text{H-NMR}$ compound (**11**) through the use (of DMSO-d₆) as a solvent. The proton signal (m, A-rH) appeared δ (6.693-7.401ppm), (s,H1) appeared δ (1.057ppm), (H2) appeared δ (1.84ppm) (d,H3) appeared δ (7.929ppm).). The $^{13}\text{CNMR}$ spectrum was used by (DMSO-d₆), showing the signal (C1) appeared δ (27.92ppm), (C2) appeared δ (21.57ppm), (C=O) appeared δ (187.86ppm), (C3) appeared δ (146.23ppm), (C4) in(-CH=) appeared δ (122.92ppm), (C-O) appeared δ (152.18ppm), (C6) appeared δ (113.36ppm), (C7) appeared δ (111.32ppm).

3-7 The compound (**15**) 2,6-bis ((E)-4-bromobenzylidene)cyclohexan-1-one is solid light yellow, yield 80.48%. The infrared spectrum of compound (**15**) showed band at (1666cm^{-1}) for (C=C) aromatic, (2927cm^{-1}) for (C-H) aliphatic, (1006cm^{-1}) for (C-Br), (1770cm^{-1}) for (C=O) Ketone, (1481cm^{-1}) bend for (-CH₂-). The $^1\text{H-NMR}$

compound (15) through the use (of DMSO-d6) as a solvent. The proton signal (m, Ar-H) appeared δ (6.91ppm), (t,H1) appeared δ (2.78ppm), (s,H2) appeared δ (1.05ppm). The ¹³CNMR spectrum was used by (DMSO-d6), showing the signal(C-Ar) appeared δ (130.2-137.41ppm), (C-Br) appeared δ (122.14ppm), (C=O) appeared δ (189.27ppm).

3-8 The compound (17) 2,6-bis((E)-4-hydroxy-3-methoxybenzylidene)cyclohexan-1-one is yellow solid, M.P (166-168), yield 72.18%. The infrared spectrum of compound (17) showed band at (3475cm⁻¹) for (OH) phenol, (2970cm⁻¹) for (C-H)aliphatic, (1517cm⁻¹) for (C=C), (1226cm⁻¹) for (C-O), (1662cm⁻¹) for (C=O). The ¹H-NMR compound (17) through the use (of DMSO-d6) as a solvent. The proton signal (s,Ar-H) appeared δ (6.092-7.052ppm), (s,OH)appeared δ (5.3ppm), (m,CH₃) appeared δ (3.36-3.38ppm), (s,H1) appeared δ (1.5ppm), (d,H2) appeared δ (1.8ppm). The ¹³CNMR spectrum was used by (DMSO-d6), showing the signal (CH₃) appeared δ (54.94ppm), (C-OH) appeared δ (117.44ppm), (C-O)appeared δ (171.33ppm), (C=O)appeared δ (85.98ppm).

BIOLOGICAL ACTIVITY

The biological activity in this research was conducted against Staphylococcus aureus, Escherichia Coli of compound (15,13), compound 13 showed the highest antibacterial potential with inhibition zone diameter of 17 mm against Escherichia Coli bacteria, while compound 15 showed antibacterial potential with inhibition zone diameter of 9 mm against Escherichia Coli bacteria. Antibacterial with a diameter of 13 mm against Staphylococcus aureus, while the other compounds showed weak inhibition against Staphylococcus aureus, Escherichia Coli.

Table.2 Biological activity of compound(C11-C17)

Antibacterial Activity				
NO. Sample	Escherichia Coli	Diameter mm	Staphylococcus aureus	Diameter mm
11				
13	++	17		
15	+	9	++	13
17				

"- =Inactive, + = (5-10) mm =slightly active, ++ = (11-20) mm middling.

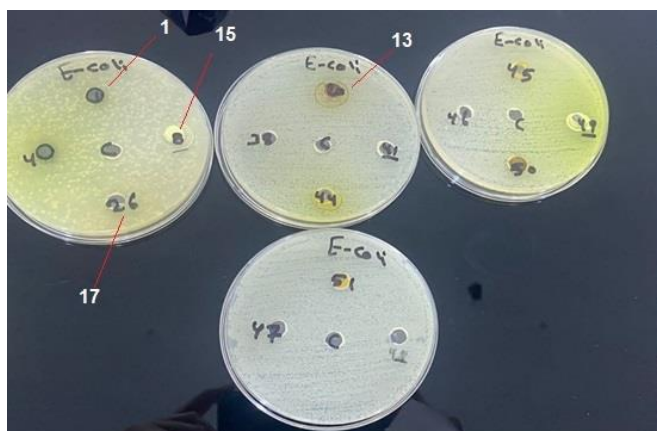


Figure. 26. Shows the effect of the compounds on the (Staphylococcus aureus, Escherichia coli)

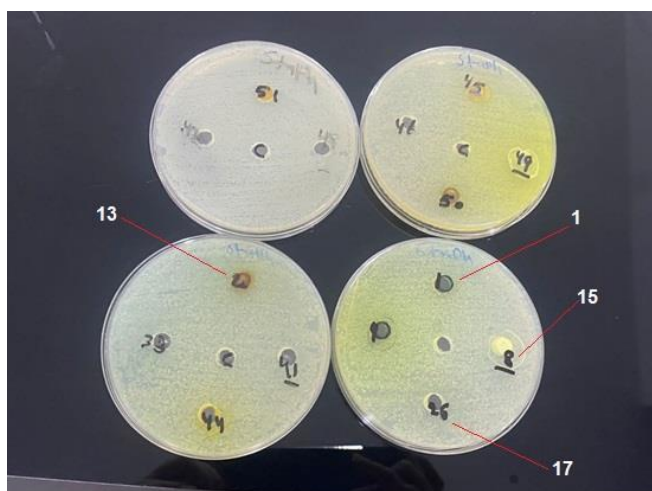


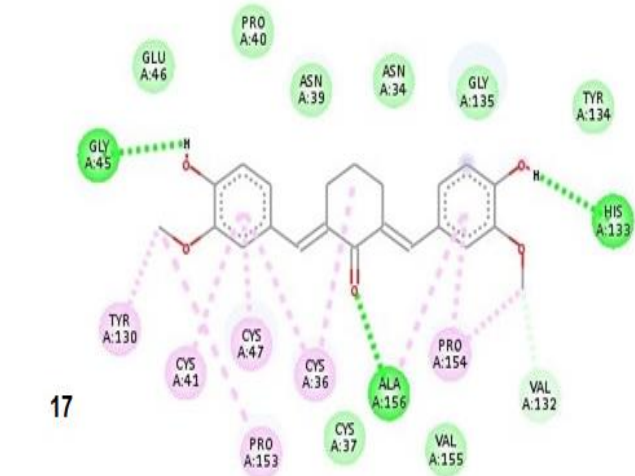
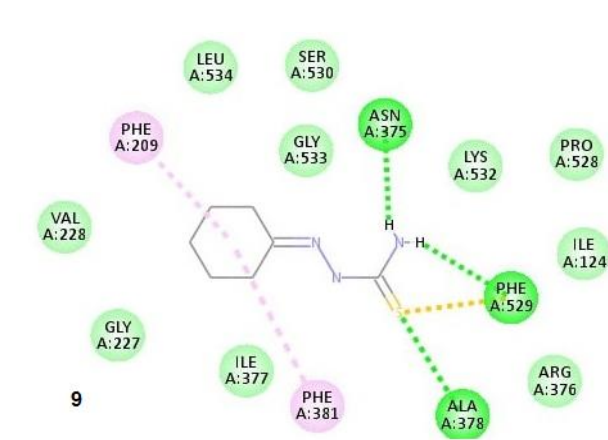
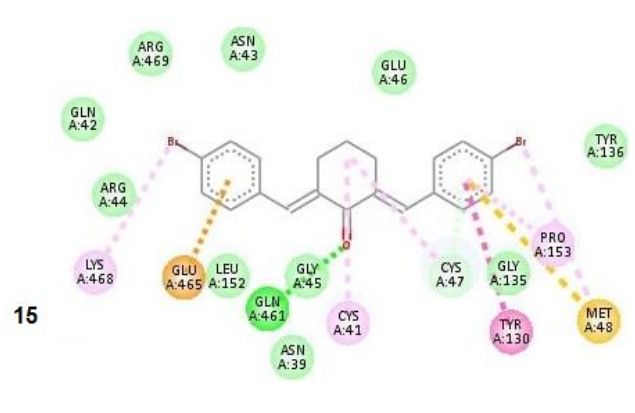
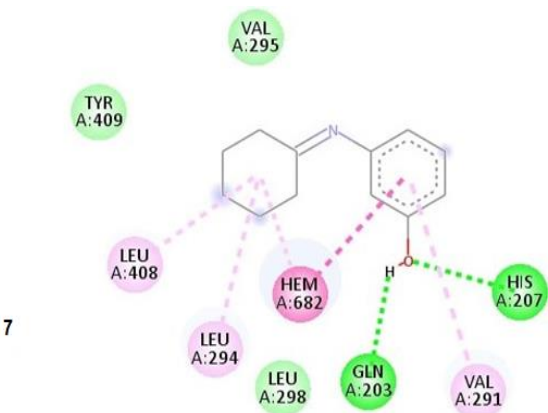
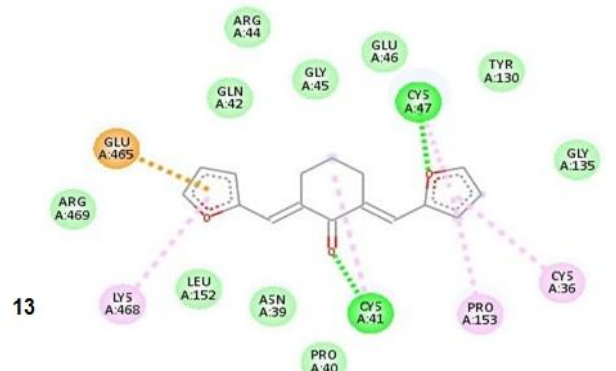
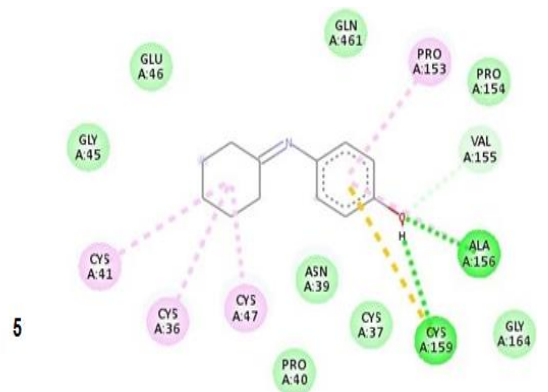
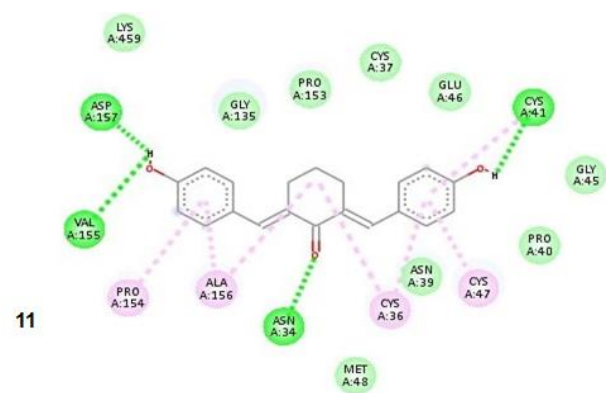
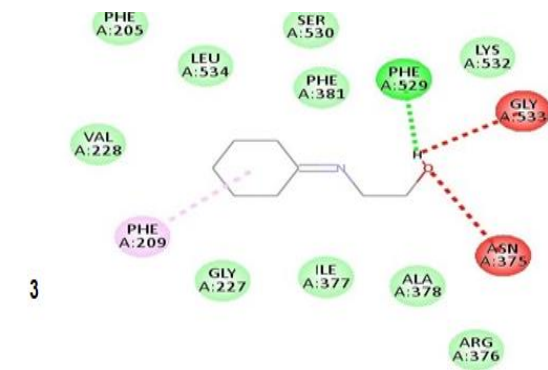
Figure. 26. Shows the effect of the compounds on the (Staphylococcus aureus, Escherichia coli)

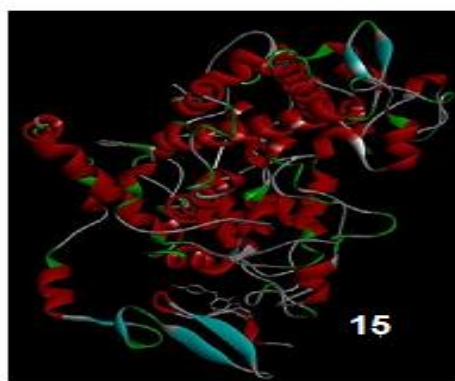
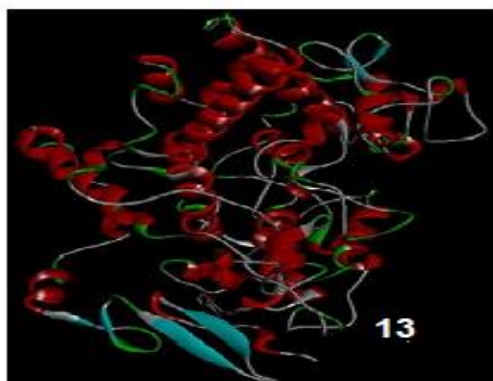
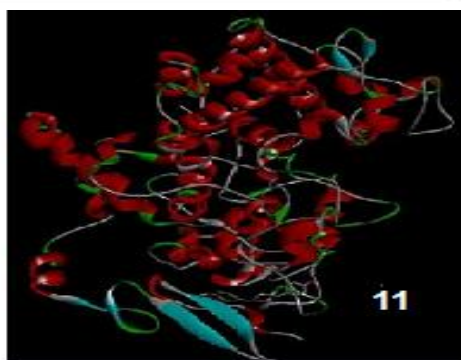
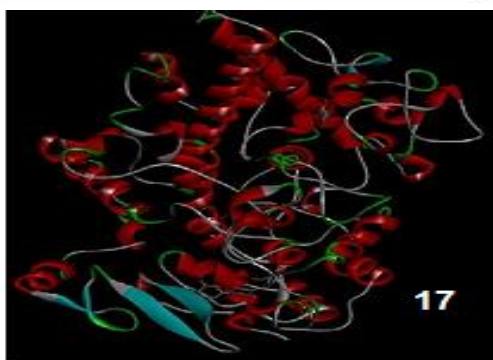
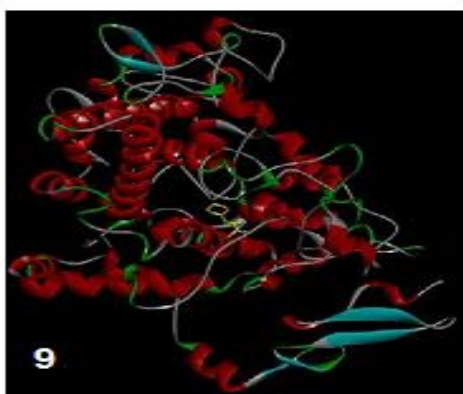
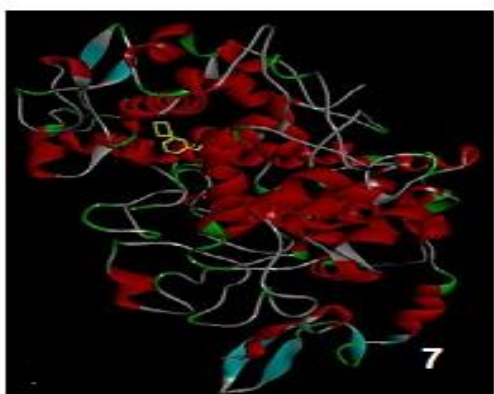
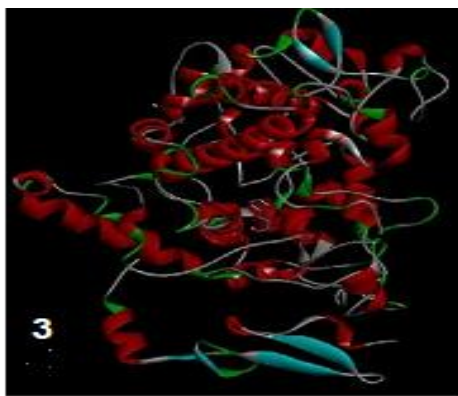
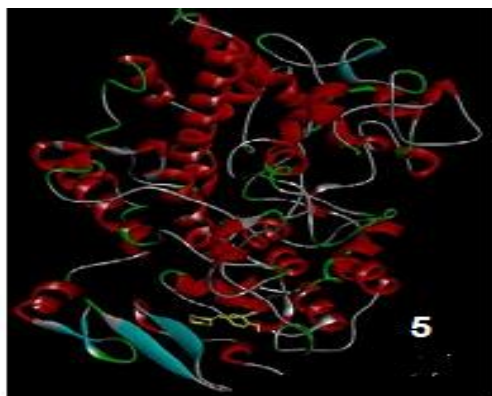
MOLECULAR DOCKING STUDY

The molecular docking of the prepared compounds with the protein responsible for breast cancer was studied using AutoDock 4.2.6 program, and the following results were obtained, shown in the table 3.

Table 3. Shows the molecular docking values of the breast cancer protein (6COX)

Compound	Lowest Binding Energy	Run
3	-5.59	50
5	-6.37	37
7	-6.90	31
9	-6.13	46
11	-8.88	18
13	-8.05	46
15	-10.13	5
17	-9.15	30





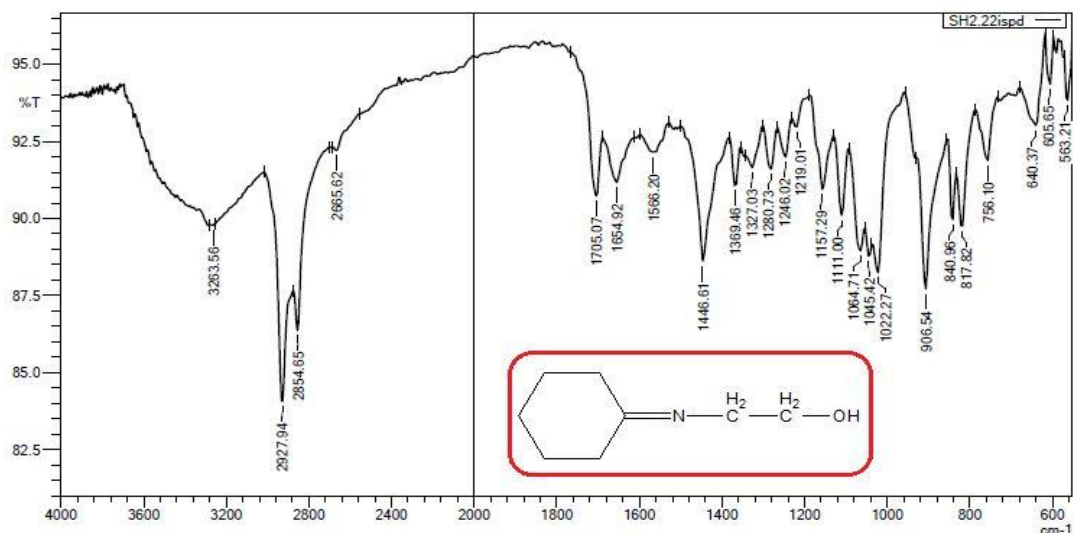
CONCLUSIONS

To summarize, I prepared a new set of Schiff bases and chalcones derived from cyclohexanone. The compounds were validated by infrared spectroscopy, ¹H-NMR and ¹³C-NMR, then the biological activity, finally, by Molecular Docking. Some compounds showed moderate biological activity against bacteria *Staphylococcus aureus*, *Escherichia coli*, while others showed weak to no activity.

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ANALYSIS OF FT-IR SPECTRA



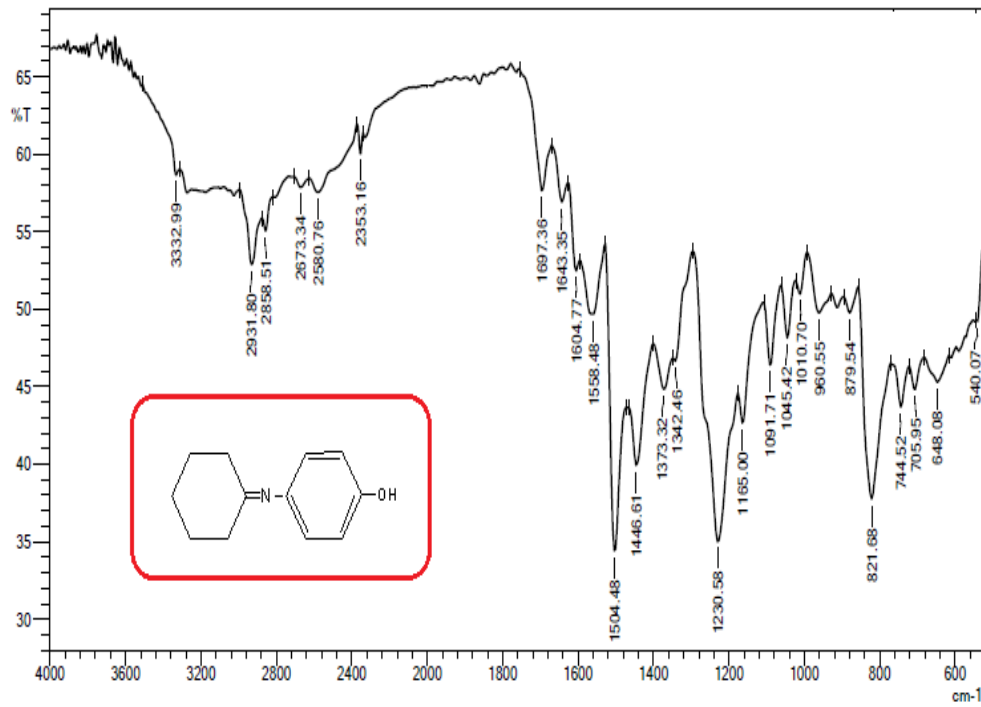


Figure 4. FT-IR spectrum of the compound (5)

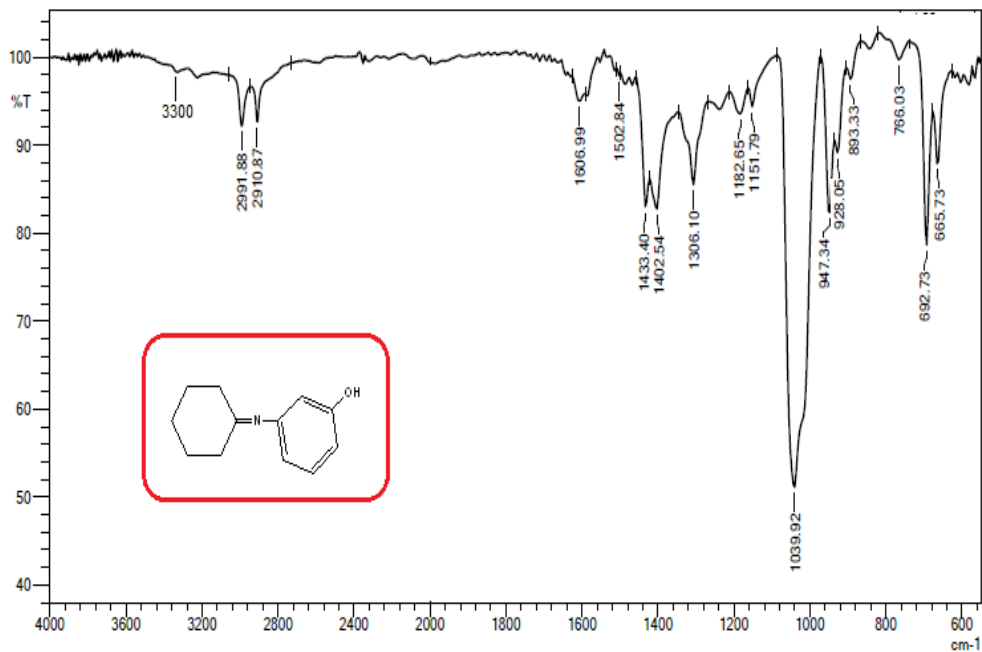


Figure 5. FT-IR spectrum of the compound (7)

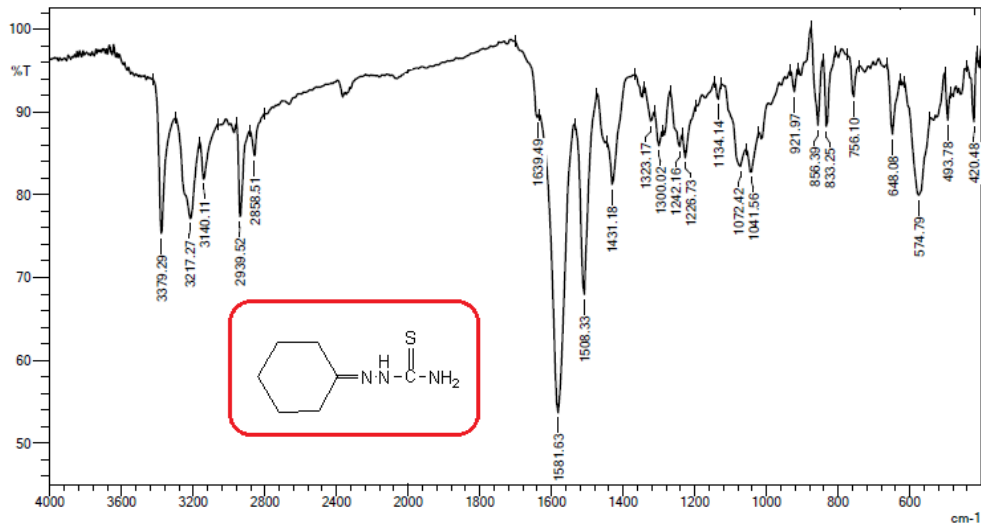


Figure 6. FT-IR spectrum of the compound (9)

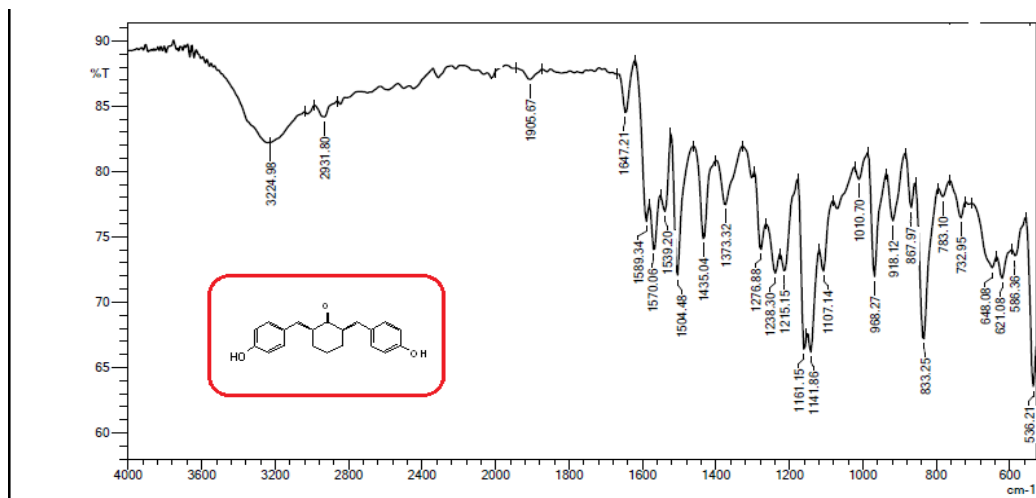


Figure 7. FT-IR spectrum of the compound (11)

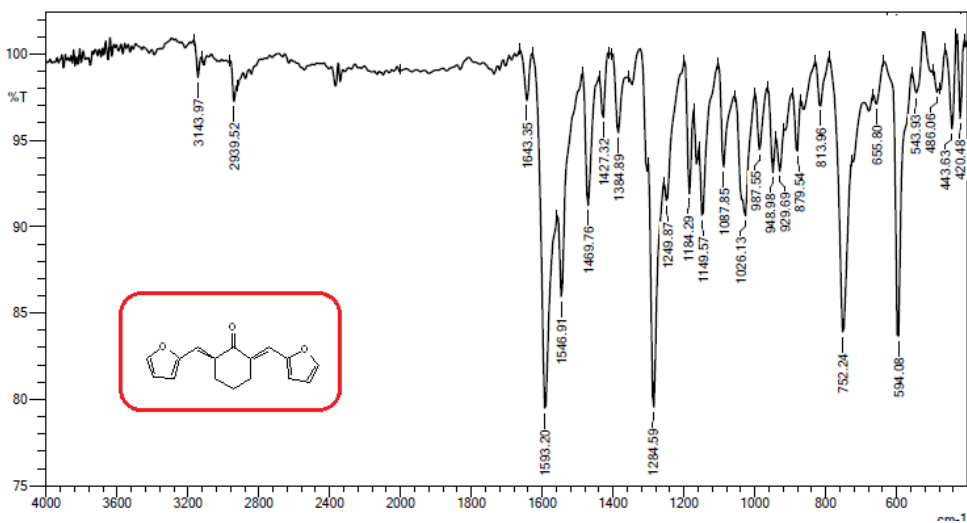


Figure 8. FT-IR spectrum of the compound (13)

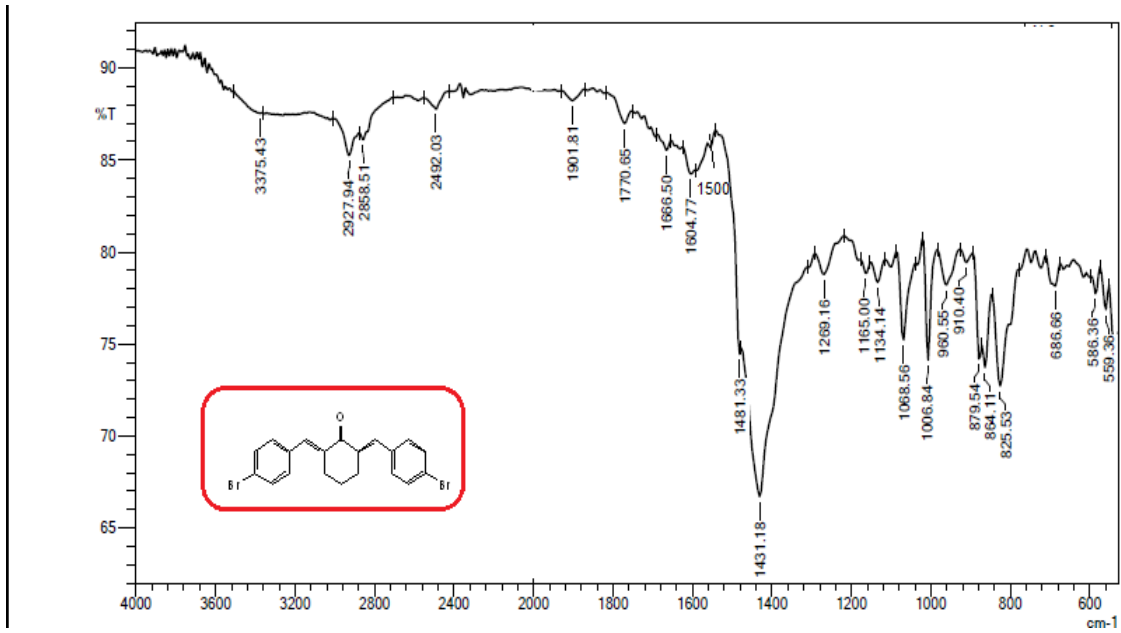


Figure 9. FT-IR spectrum of the compound (15)

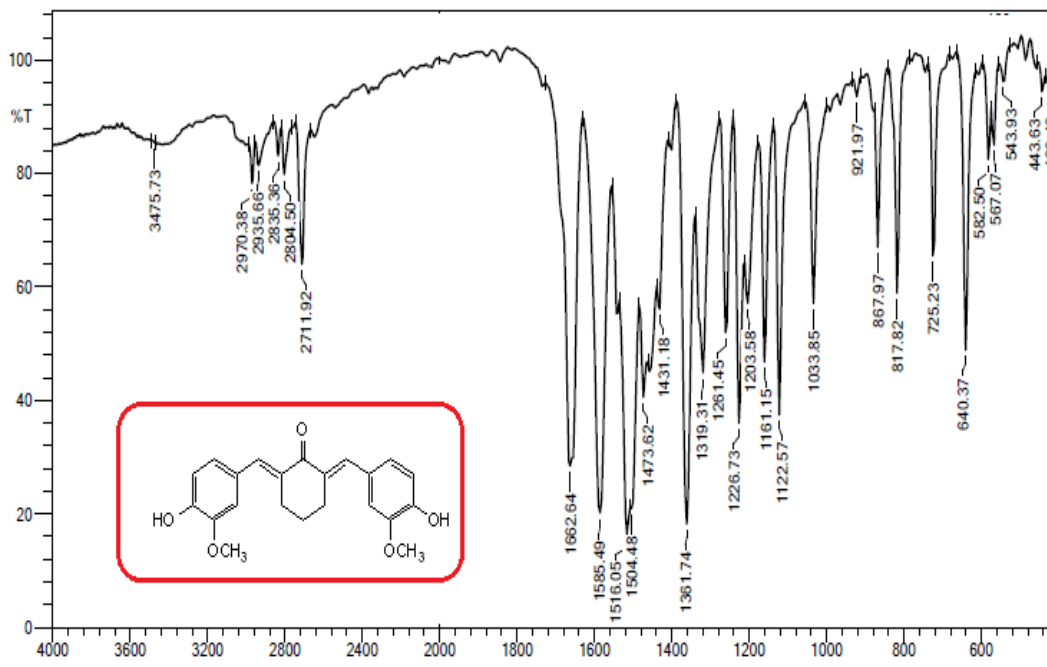


Figure 10. FT-IR spectrum of the compound (17)

¹HNMR and ¹³C NMR spectrum

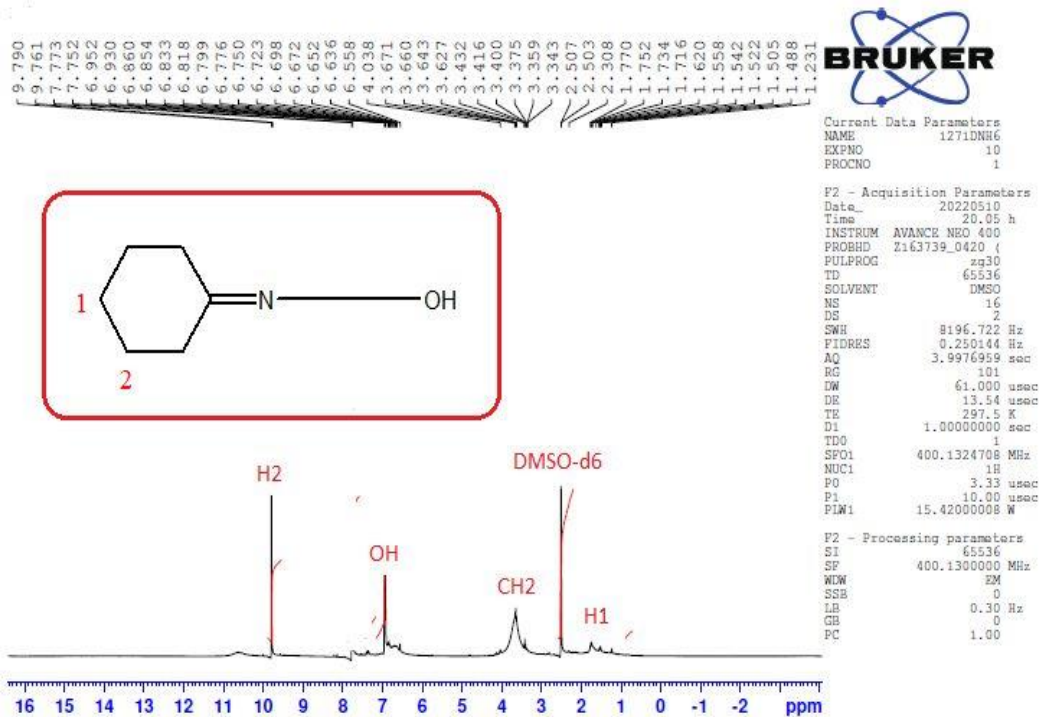


Figure 11. ¹HNMR spectrum of the compound (3)

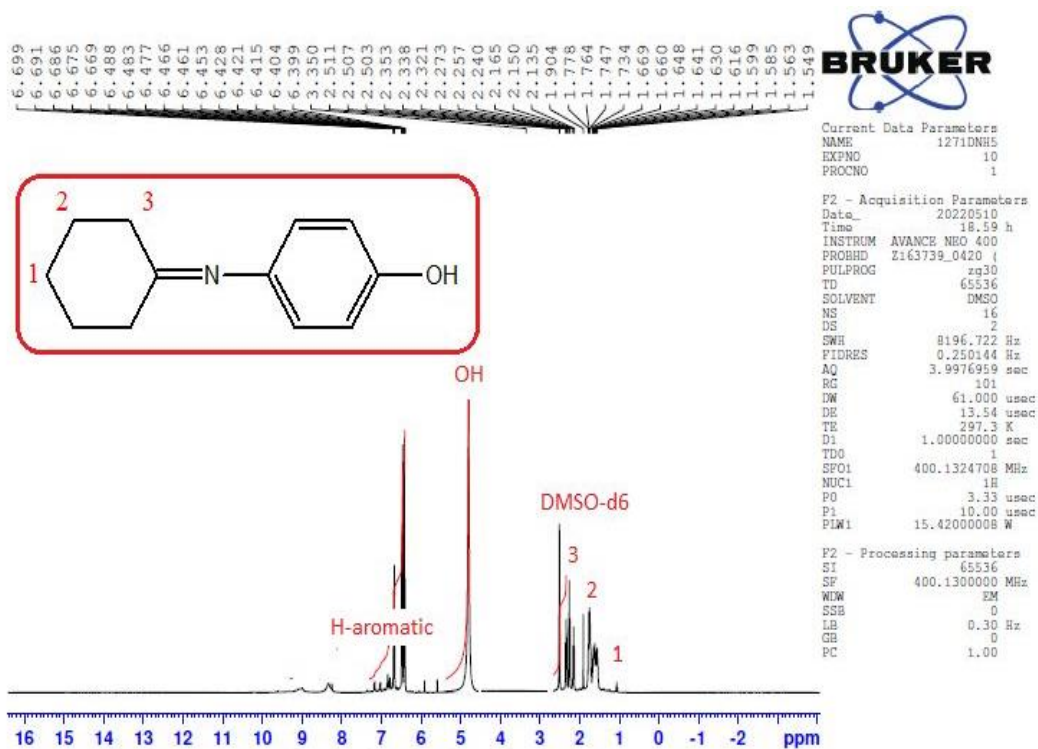


Figure 12. ¹HNMR spectrum of the compound (5)

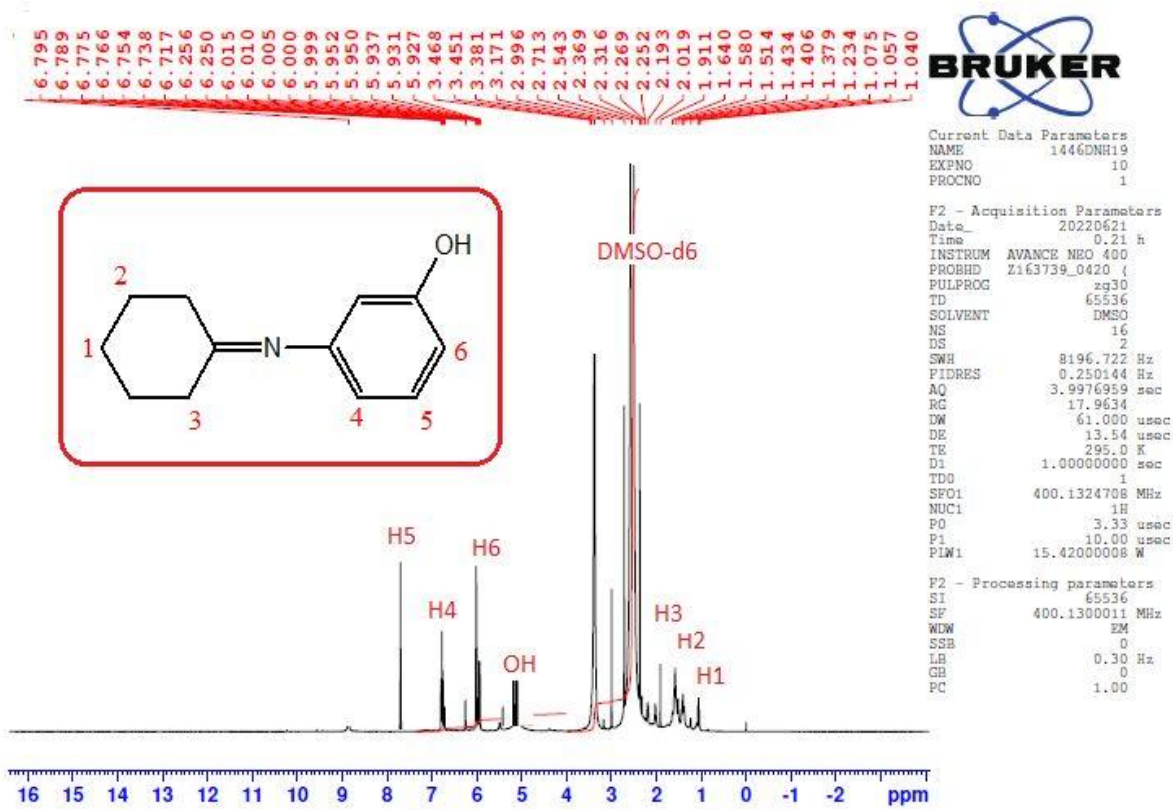


Figure 13. ¹H NMR spectrum of the compound (7)

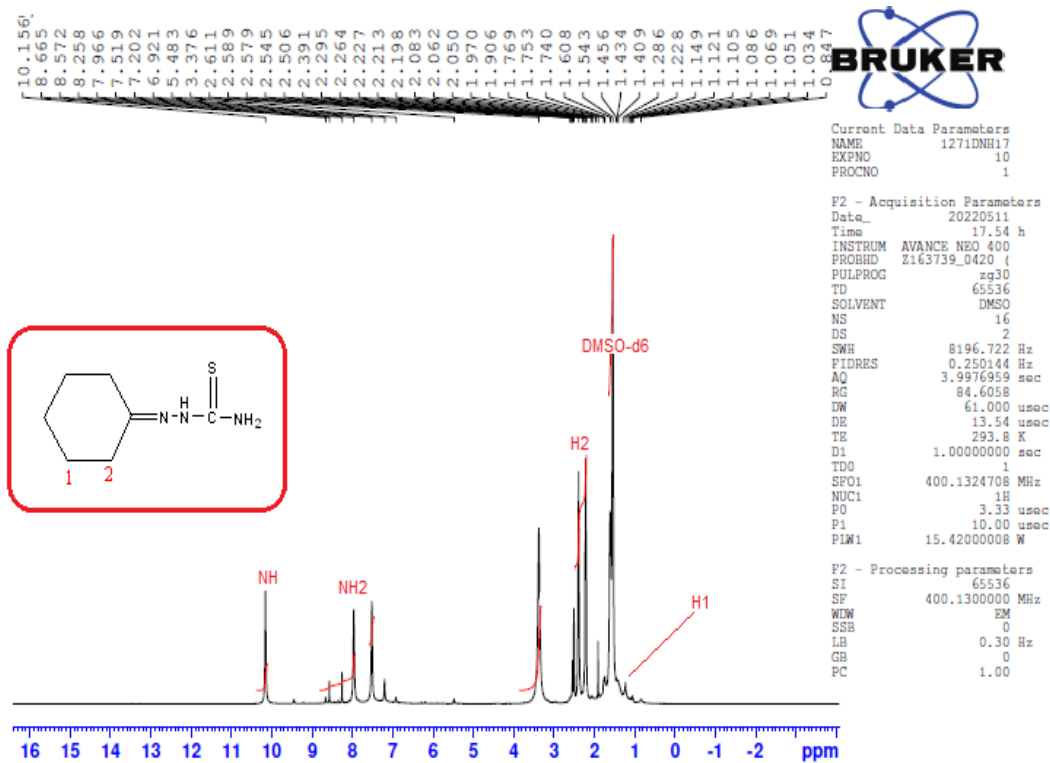


Figure 14. ¹H NMR spectrum of the compound (9)

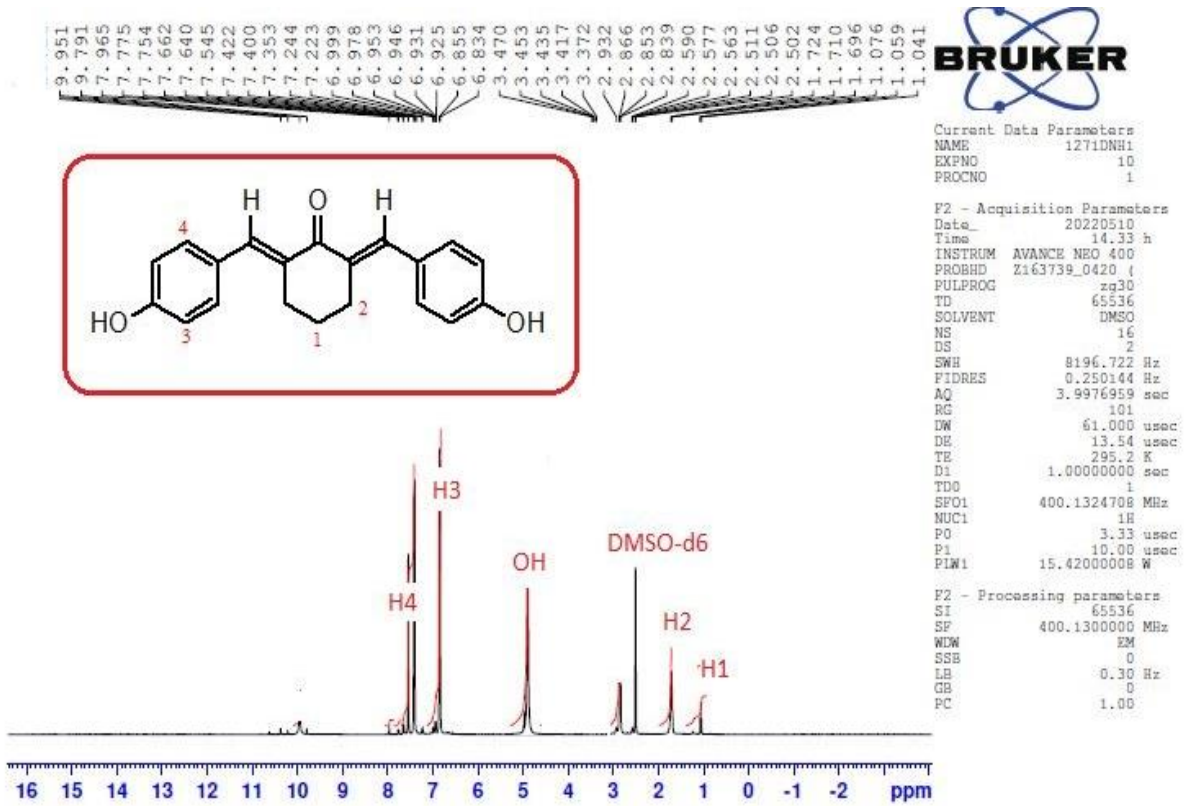


Figure 15. ¹H NMR spectrum of the compound (11)

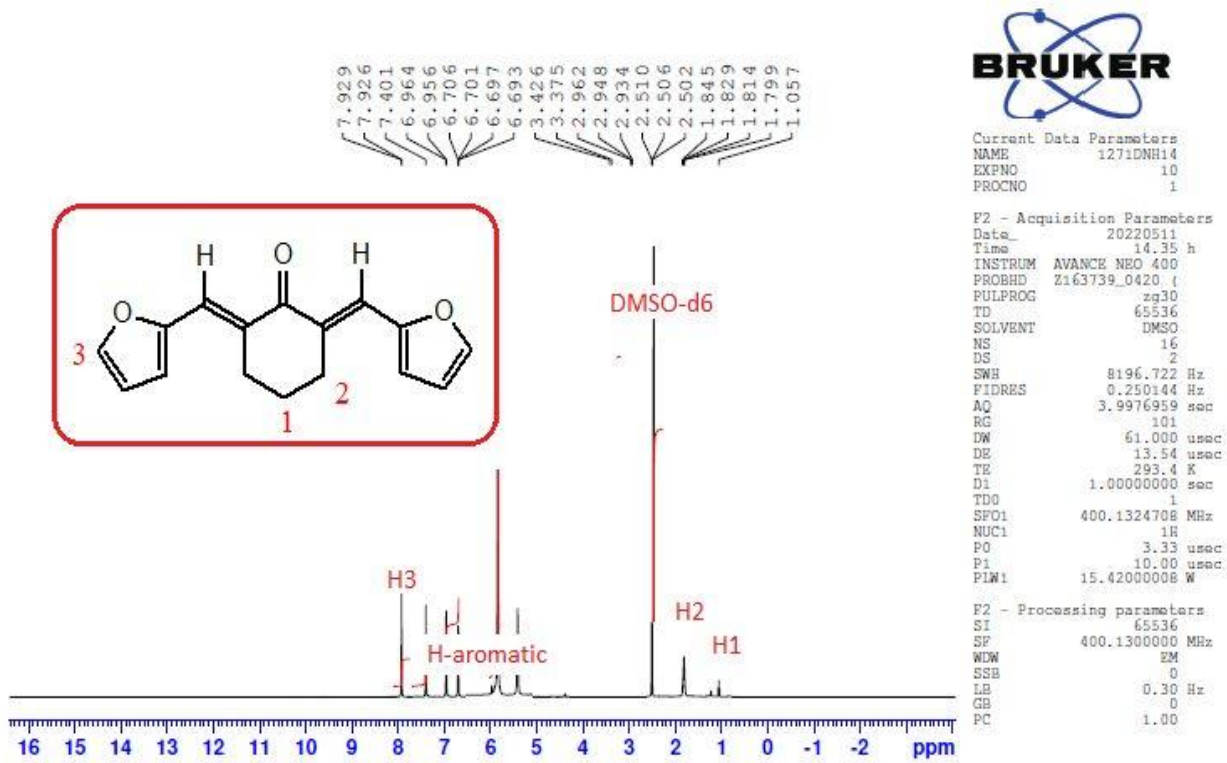


Figure 16. ¹H NMR spectrum of the compound (13)

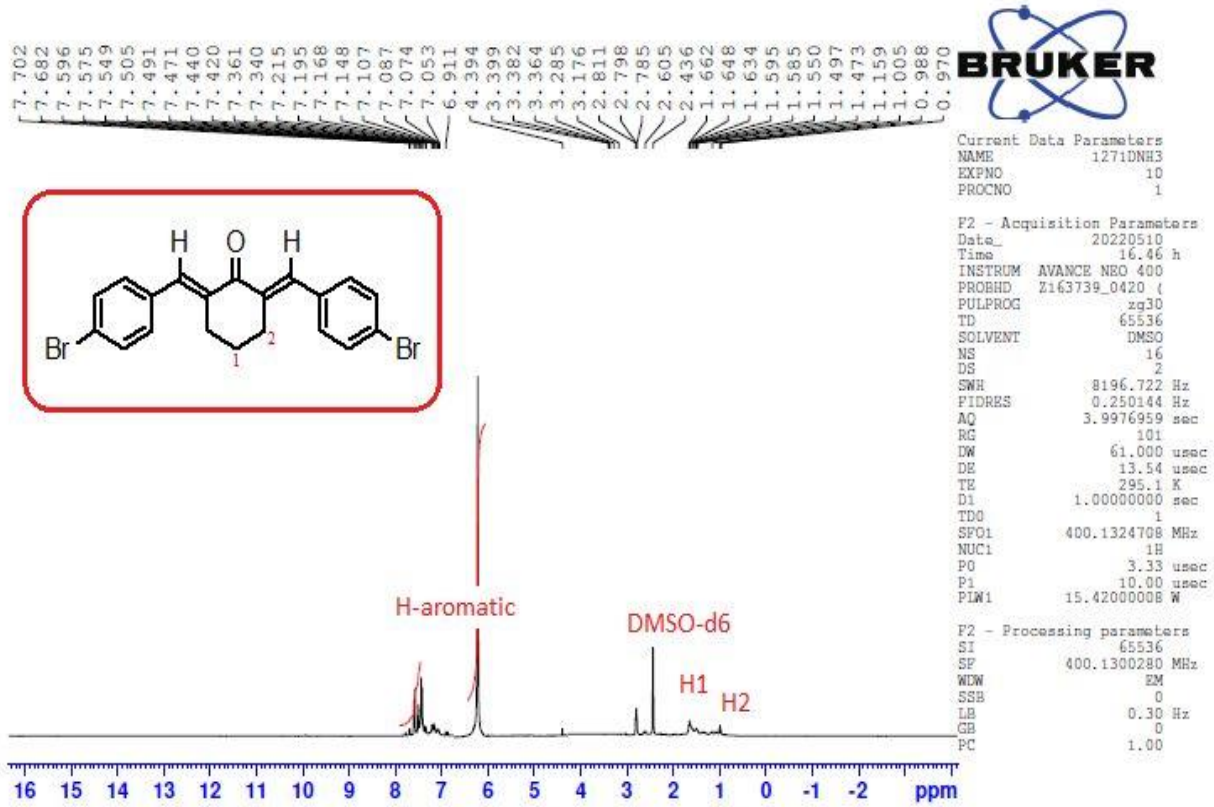


Figure 17. ¹HNMR spectrum of the compound (15)

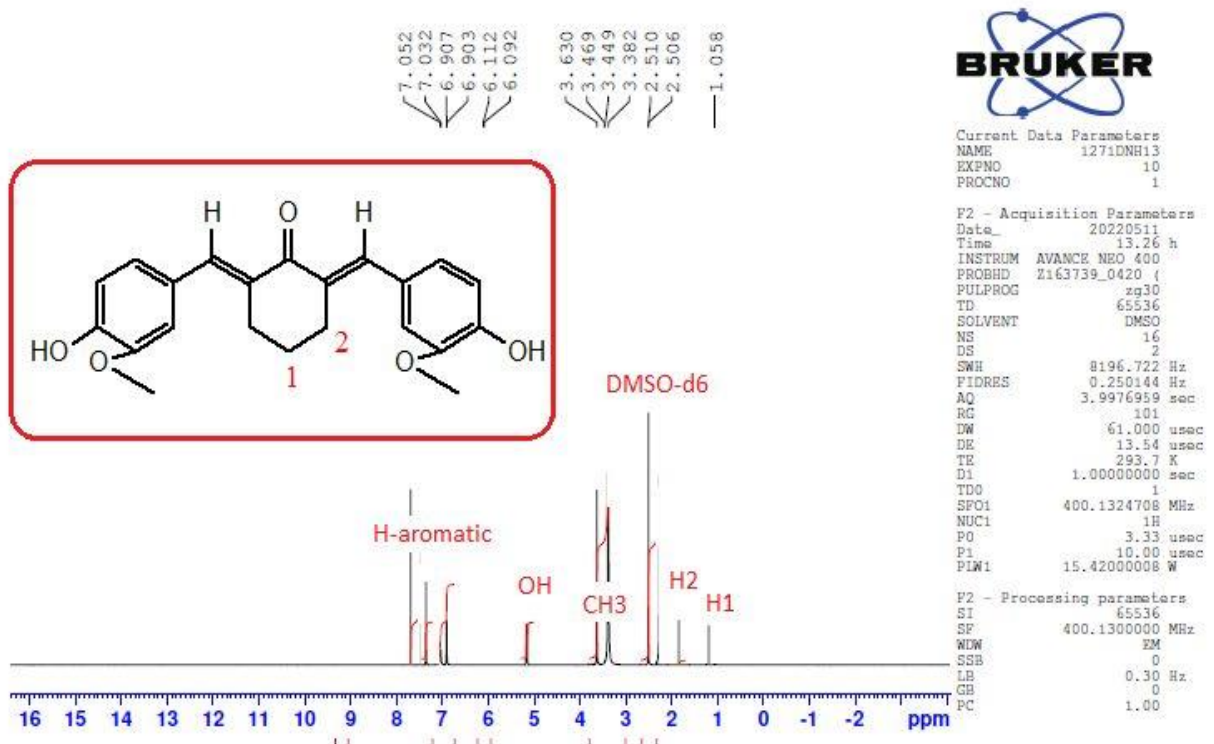


Figure 18. ¹HNMR spectrum of the compound (17)

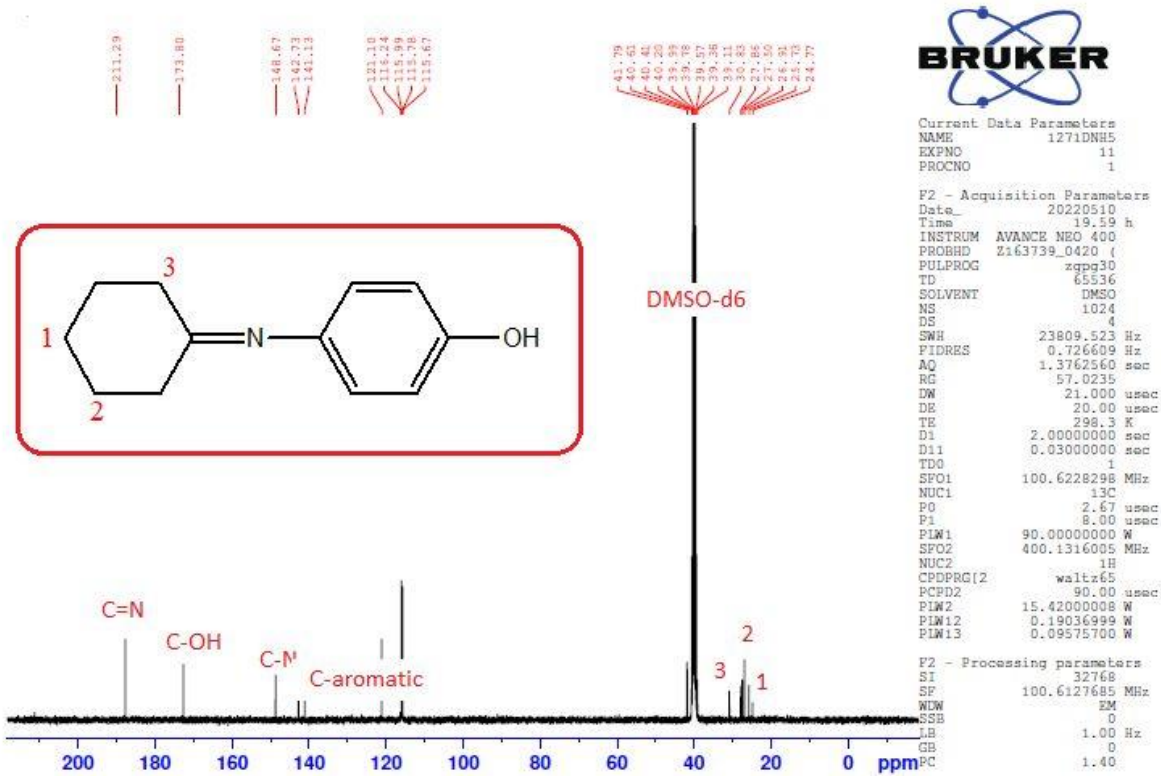


Figure 19. ¹³C NMR spectrum of the compound (5)

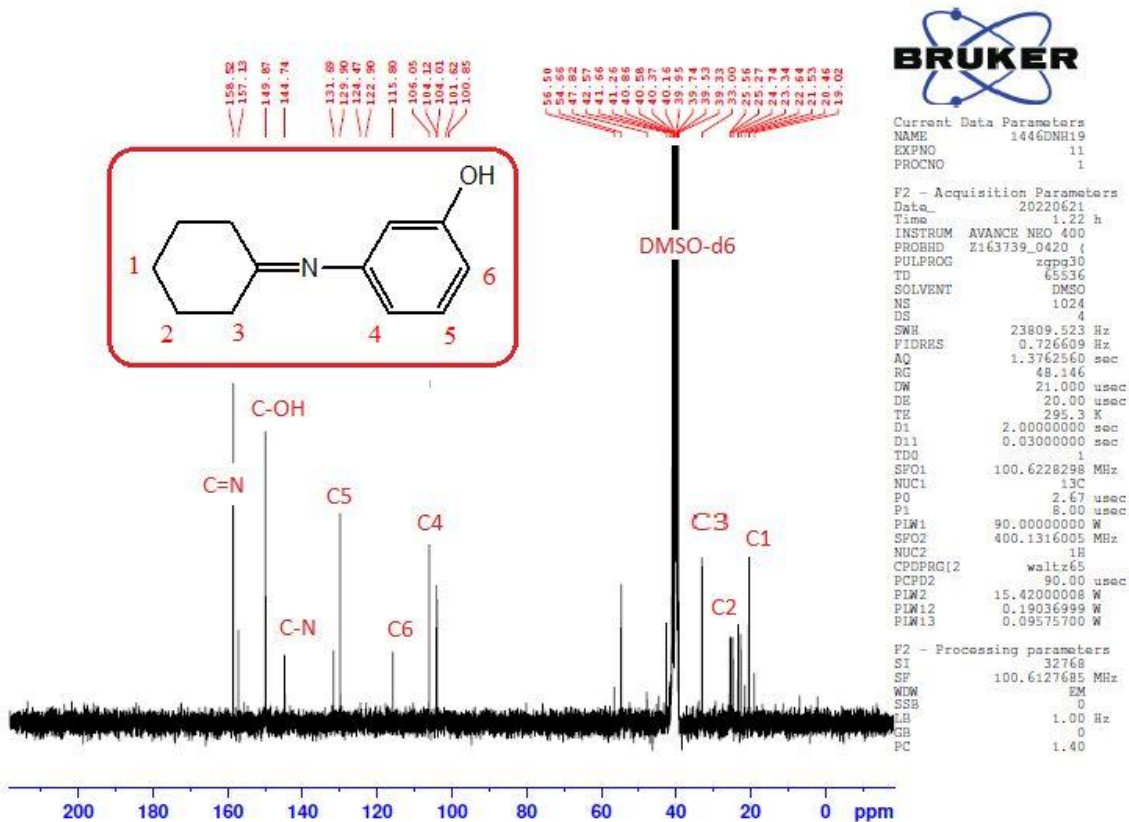


Figure 20. ¹³C NMR spectrum of the compound (7)

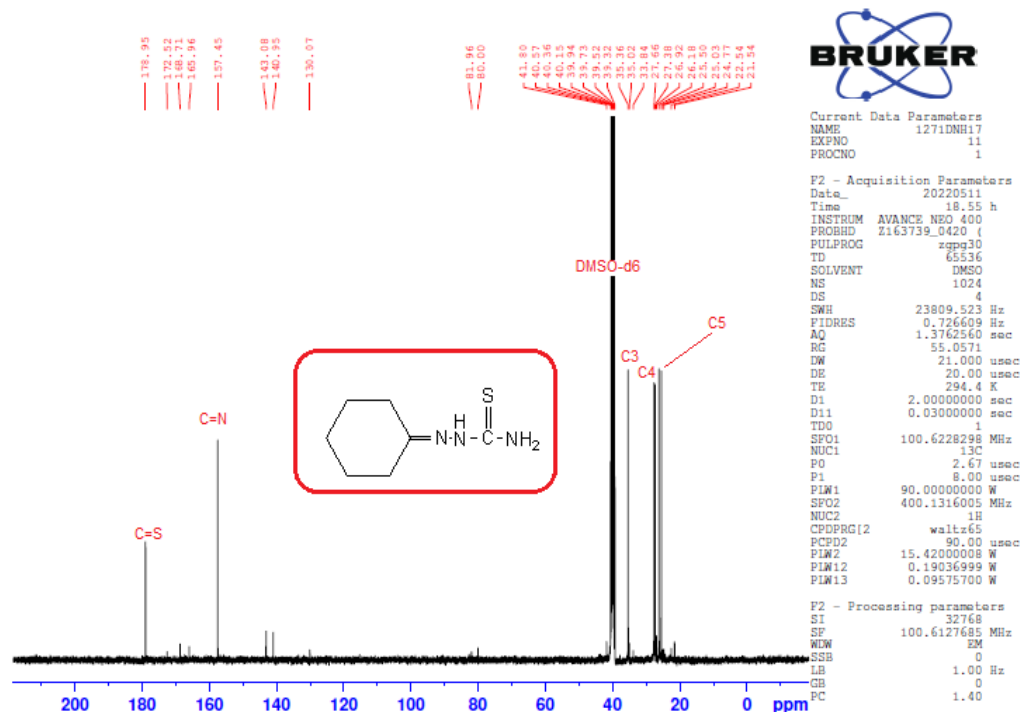


Figure 21. ¹³C NMR spectrum of the compound (9)

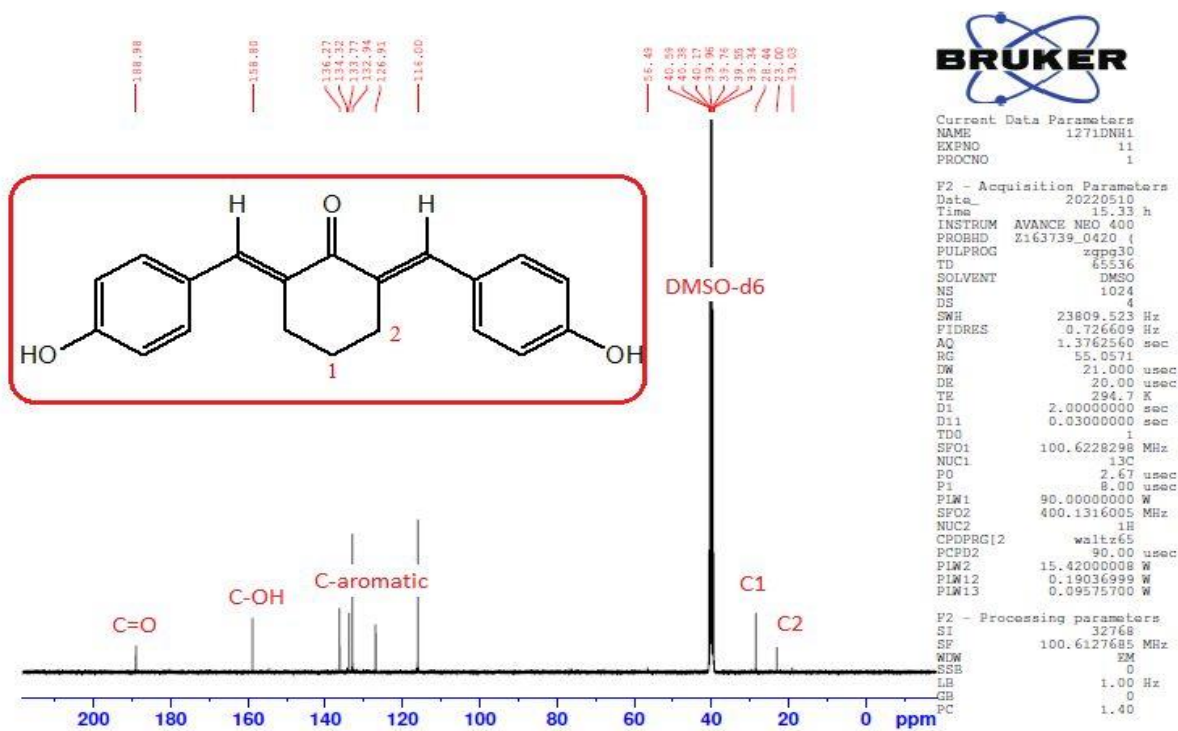


Figure 22. ¹³C NMR spectrum of the compound (11)

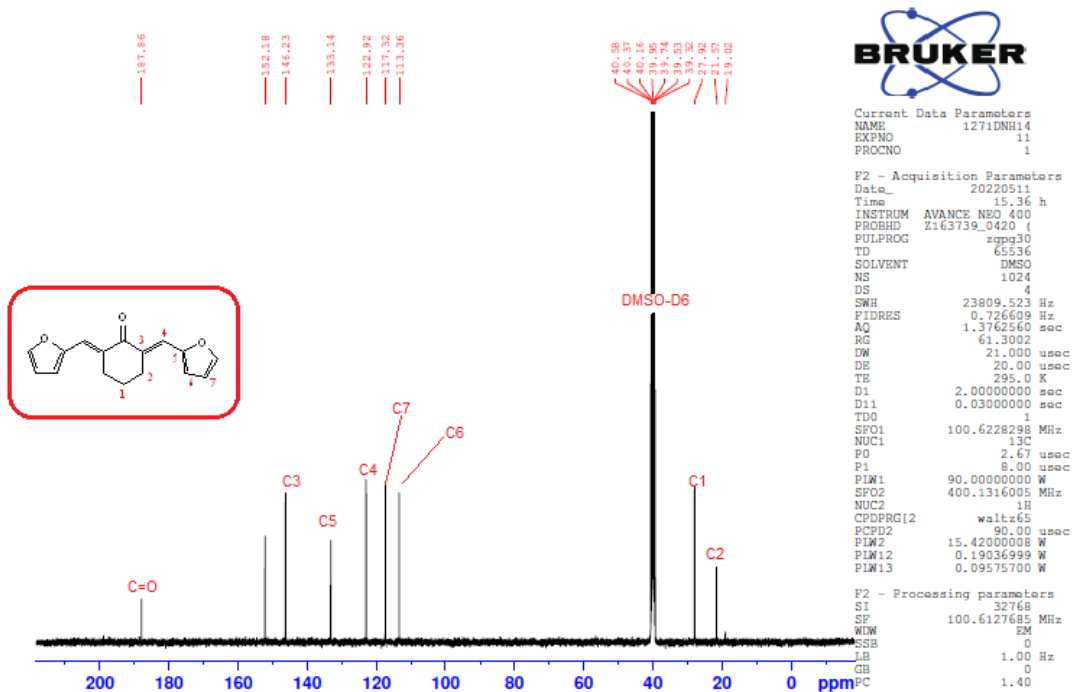


Figure 23. ¹³C NMR spectrum of the compound (13)

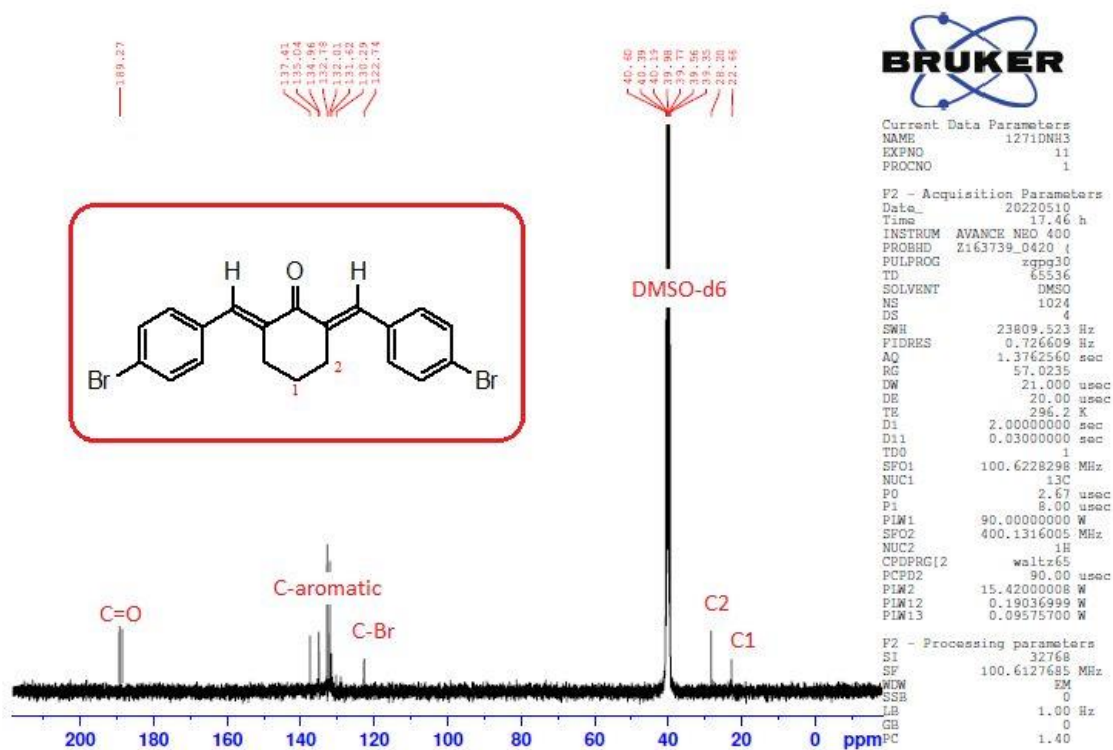


Figure 24. ¹³C NMR spectrum of the compound (15)

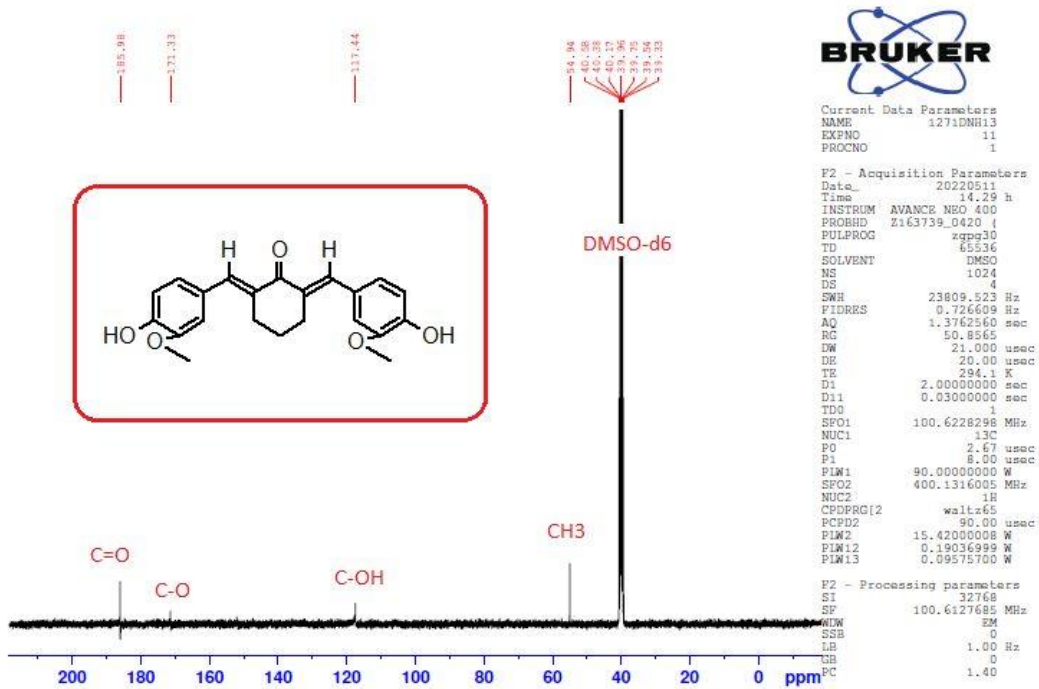


Figure 25. ¹³C NMR spectrum of the compound (17)